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ELECTROCHEMICAL AND OPTICAL ELECTRON TRANSFER PROCESSES 1/1

(U) NEW YORK UNIV NY DEPT OF CHEMISTRY P DELAHAY

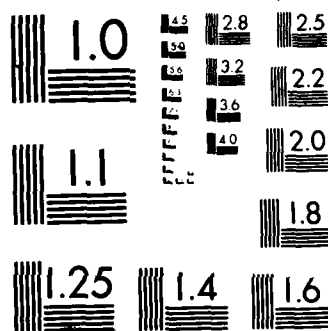
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PHOTOCOPY RESOLUTION TEST CHART

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FINAL REPORT

12

ELECTROCHEMICAL AND OPTICAL ELECTRON TRANSFER PROCESSES

Paul Delahay  
Principal Investigator

Period: April 1, 1982 to September 30, 1986

New York University  
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New York, NY

September 30, 1986

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FIELD	GROUP	SUB-GROUP	Dielectric dispersion      Ionization energies		
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19. ABSTRACT (Continue on reverse if necessary and identify by block number) A summary of significant results is given covering the three broad areas of investigation: (1) Nonequilibrium electronic polarization and loss in optical electron transfer: (a) first observation of this effect and subsequent detailed experimental study; (b) development of a detailed theory accounting for experimental results. (2) Inner-sphere nuclear reorganization in optical electron transfer: (a) treatment of cations and metal complexes and correlation between optical electron transfer and thermal electron exchange; (b) development of the theoretical solvation model of inner-sphere reorganization for univalent anions and experimental verification of this model; (c) application to the energetics of anion/radical couples in aqueous solution; (d) application to the gas-liquid shift for photoelectron emission. (3) Ionization energies of liquids from energy distribution, quantum yield and second derivative curves. A list of reports and publications arising from this work is supplied.					
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## SUMMARY OF SIGNIFICANT RESULTS

Three broad areas were covered:

(1) Nonequilibrium electronic polarization and loss in optical electron transfer

- (a) First observation of this effect and subsequent detailed experimental study (report 3)
- (b) Development of a detailed theory accounting for experimental results (reports 3, 6, 8 and 11)

Yield spectra for photoelectron emission by liquids and solutions exhibit a fine structure which is essentially determined by the nature of the solvent and not by the species being photoionized. This fine structure was shown to arise from nonequilibrium electronic processes (polarization, loss) arising from dielectric dispersion of the solvent. Theoretical and experimental fine structure features agree extremely well. This effect, which was first observed in our laboratory, is the electronic counterpart of nuclear reorganization (Marcus, Sutin, etc.)

(2) Inner-sphere nuclear reorganization in optical electron transfer

- (a) Treatment of cations and metal complexes and correlation between optical electron transfer and thermal electron exchange (report 4)
- (b) Development of the theoretical solvation model of inner-sphere reorganization for univalent anions and experimental verification of this model (reports 4 and 9)
- (c) Application to the energetics of anion/radical couples in aqueous solution (report 10)
- (d) Application to the gas-liquid shift for photoelectron emission (report 5)

Energies of inner-sphere reorganization were derived from a bond stretching model for optical electron transfer involving cations and metal complexes. This made it possible to correlate thermal and optical electron transfer processes. Results are in agreement with experiment. The energy of inner-sphere reorganization of univalent anions was calculated from a solvation model based on a multipole expansion accounting for ion-solvent electrostatic interactions. Other contributions (London dispersion, Born repulsion, etc.) were also taken into account. Agreement with experiment is achieved.

(3) Ionization energies of liquids from energy distribution, quantum yield and second derivative curves

These three methods of determining ionization energies were investigated and compared for eight liquids (report 7).

TECHNICAL REPORTS AND PUBLICATIONS

1. P. Delahay, "Anomalous reorganization free energies in optical electron transfer in solution," Technical Report No. 1 (June 1982); Chem. Phys. Lett. 90, 425 (1982).
2. P. Delahay, "Dielectric dispersion in optical electron transfer in solution," Technical Report No. 2 (March 1983); Chem. Phys. Lett. 96, 613 (1983).
3. P. Delahay and A. Dziedzic, "Dispersion spectroscopy of optical electron transfer in solution," Technical Report No. 3 (February 1984); J. Chem. Phys. 80, 5381 (1984).
4. P. Delahay and A. Dziedzic, "Inner-sphere reorganization in optical electron transfer," Technical Report No. 4 (March 1984); J. Chem. Phys. 80, 5793 (1984).
5. P. Delahay and A. Dziedzic, "Gas-liquid correlation of ionization energies," Technical Report No. 5 (April 1984); Chem. Phys. Lett. 108, 169 (1984).
6. P. Delahay and A. Dziedzic, "Solvation and dielectric dispersion in optical electron transfer," Technical Report No. 6 (July 1984); J. Chem. Phys. 81, 3678 (1984).
7. K. P. Cheung, I. Watanabe, A. Dziedzic, K. von Burg and P. Delahay, "Ionization energies of liquids from energy distribution, quantum yield and second derivative curves," Technical Report No. 7 (March 1985); J. Electron Spectrosc. 36, 245 (1985).
8. P. Delahay and A. Dziedzic, "Nonequilibrium electronic polarization of the solvent in photoionization," Technical Report No. 8 (July 1985); J. Chem. Phys. 84, 936 (1986).
9. P. Delahay and A. Dziedzic, "Discrete model for inner-sphere reorganization of anions," Technical Report No. 9 (May 1986); Chem. Phys. Lett., 128, 378 (1986).
10. P. Delahay and A. Dziedzic, "Nuclear reorganization in the photoionization of anions in solution," Technical Report No. 10 (July 1986); Proc. Indian Acad. Sci. (Chem. Sci.), in press (by invitation from the editor; issue in honor of K. S. G. Doss).
11. P. Delahay and A. Dziedzic, "Transition dipole-solvent interaction in photoionization," Technical Report No. 11 (September 1986); Chem. Phys. Lett., in press.
12. P. Delahay and A. Dziedzic, "Transition dipole-solvent interaction in photoionization in solution," Technical Report No. 12 (September 1986); J. Electroanal. Chem., in press (by invitation from the editor; special issue

in honor of H. Gerischer on his retirement as Director of the Fritz-Haber-Institut, Berlin).

13. In addition to the above reports and publications, the principal investigator wrote Chapter 2 on photoelectron emission spectroscopy of liquids and solutions in "Electron Spectroscopy," vol. 5, C. R. Bundle and A. D. Baker, editors (Academic Press, London, 1984), pp. 123-196.

#### PERSONNEL

Charles Cheung, graduate student, Ph.D. degree (now with ATT-Bell Laboratories)

Andrew Dziedzic, graduate student, Ph.D. degree, and postdoctoral fellow (now with Goldman Sachs)



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